

Phys 221 (Section 8)

Quiz #6

1. A pipe which is closed at one end is excited at a frequency of 392 Hz and gives its first resonance (the fundamental) when its length is 22.2 cm. From this data, what is the speed of sound in the air column of the tube?

For the first harmonic in a tube closed at one end, we have

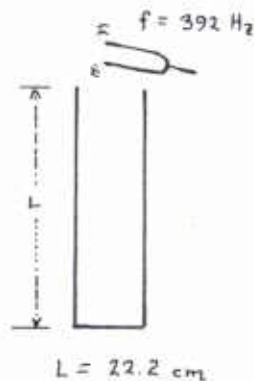
$$L = \frac{\lambda}{4} \rightarrow \lambda = 4L$$

Then

$$f = \frac{v}{\lambda} = \frac{v}{4L} \quad \text{So} \quad v = 4Lf = \text{speed of sound}$$

Substitute values given:

$$v = 4(0.222 \text{ m})(392 \text{ s}^{-1}) = \boxed{348 \frac{\text{m}}{\text{s}}}$$



2. An ideal gas at a pressure of 0.5 atm, temperature 27.0° C and volume 2.0 L is compressed to a final volume of 0.6 L and pressure of 3.0 atm. What is the final temperature of the gas?

Ideal gas law: $PV = nRT$. If n remains constant, then

$$\frac{PV}{T} = \text{constant} \quad \text{So} \quad \frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \text{for any two sets of therm. variables.}$$

Then, with $T_1 = (27.0 + 273.15) \text{ K} = 300 \text{ K}$,

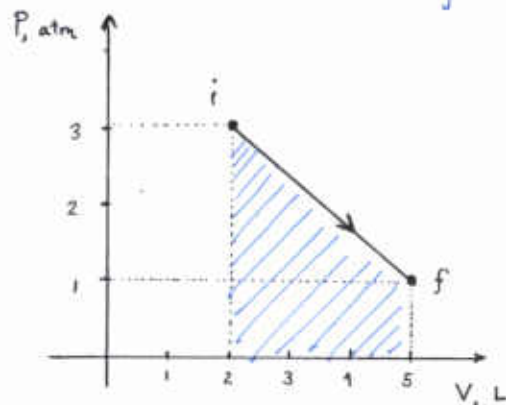
$$T_2 = \frac{P_2 V_2}{P_1 V_1} T_1 = \frac{(3.0 \text{ atm})(0.6 \text{ L})}{(0.5 \text{ L})(2.0 \text{ L})} (300 \text{ K}) = \boxed{540 \text{ K}} \quad \left[\begin{array}{l} \text{In Celsius,} \\ T_2 = (540 - 273.15)^\circ \text{C} \\ = 267^\circ \text{C} \end{array} \right]$$

3. A gas expands from i to f along the path indicated in this P - V diagram. Calculate the work done by the gas; express the answer in Joules.

Work done by the gas is $W = \int P dV$. This is the "area under the curve" for the P vs. V plot of the gas expansion.

Simple geometry gives the area shown:

$$\begin{aligned} W = \int P dV &= (1 \text{ atm})(3 \text{ L}) + \frac{1}{2}(2 \text{ atm})(3 \text{ L}) \\ &= 6.0 \text{ L} \cdot \text{atm} \\ &= (6.0 \text{ L} \cdot \text{atm}) \left(\frac{101.3 \text{ J}}{1 \text{ L} \cdot \text{atm}} \right) = \boxed{608 \text{ J}} \end{aligned}$$



$$1.00 \text{ L} \cdot \text{atm} = 101.3 \text{ J}$$

$$R = 0.0821 \frac{\text{L} \cdot \text{atm}}{\text{mol} \cdot \text{K}}$$

$$N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$$

$$T_C = T - 273.15$$